







Webinar Ghent University: Electrochemical methods for metal recovery.

# Selective Capacitive Deionization (CDI): An innovative method for metal recovery.





### 21/03/2024 Adrián Delgado Béjar

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Water stress is been always a concern for researchers, leading to new methods that explore all possible ways to meet this water demand.

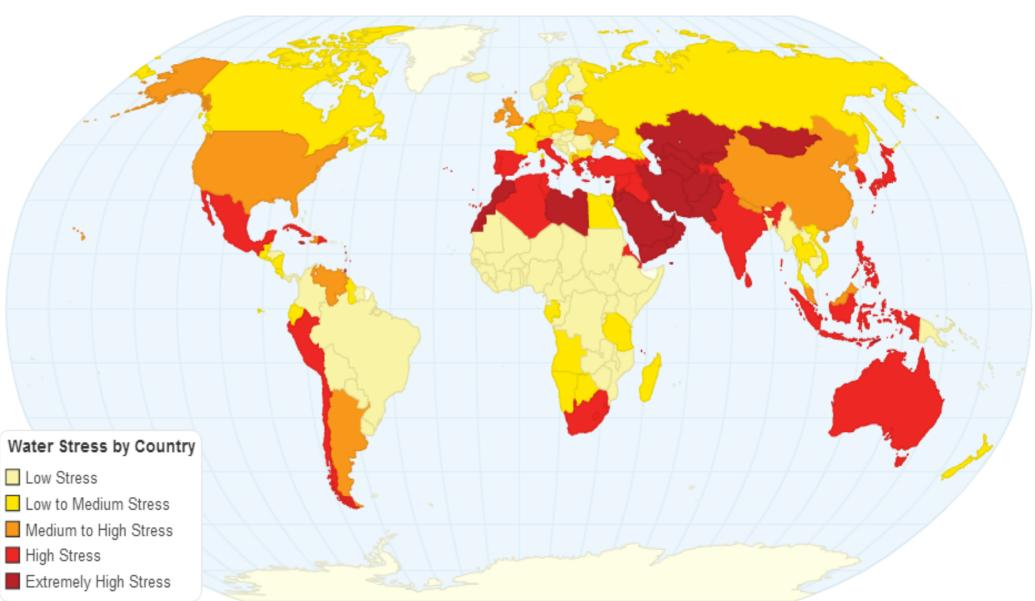
**Seawater** has been targeted due to its extensive availability.

Countries are seeking for more cost effective approaches for desalination of water:

➡ Current technologies require big amount of energy.





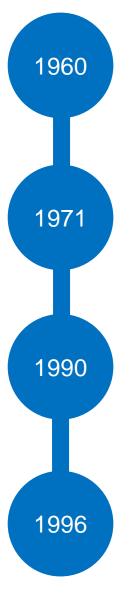




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# **Origin of Capacitive Deionization**



Electrochemical demineralization was reported: ions removed by electrochemical reactions with chemical groups of carbon particles of the electrodes – Blair and Murphy.

Johnson and Newman introduced theory for ion transport in porous carbon electrodes for ion storage according to a capacitor mechanism.

The technology attracted more attention because of the development of new electrode materials.

Farmer et al. introduced the term Capacitive Deionization and used the now commonly abbreviation CDI for the first time.







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Capacitive Deionization as a **better alternative** for these existing technologies:

- energy efficiency.
- environmental friendliness.
- ion removal efficiency.



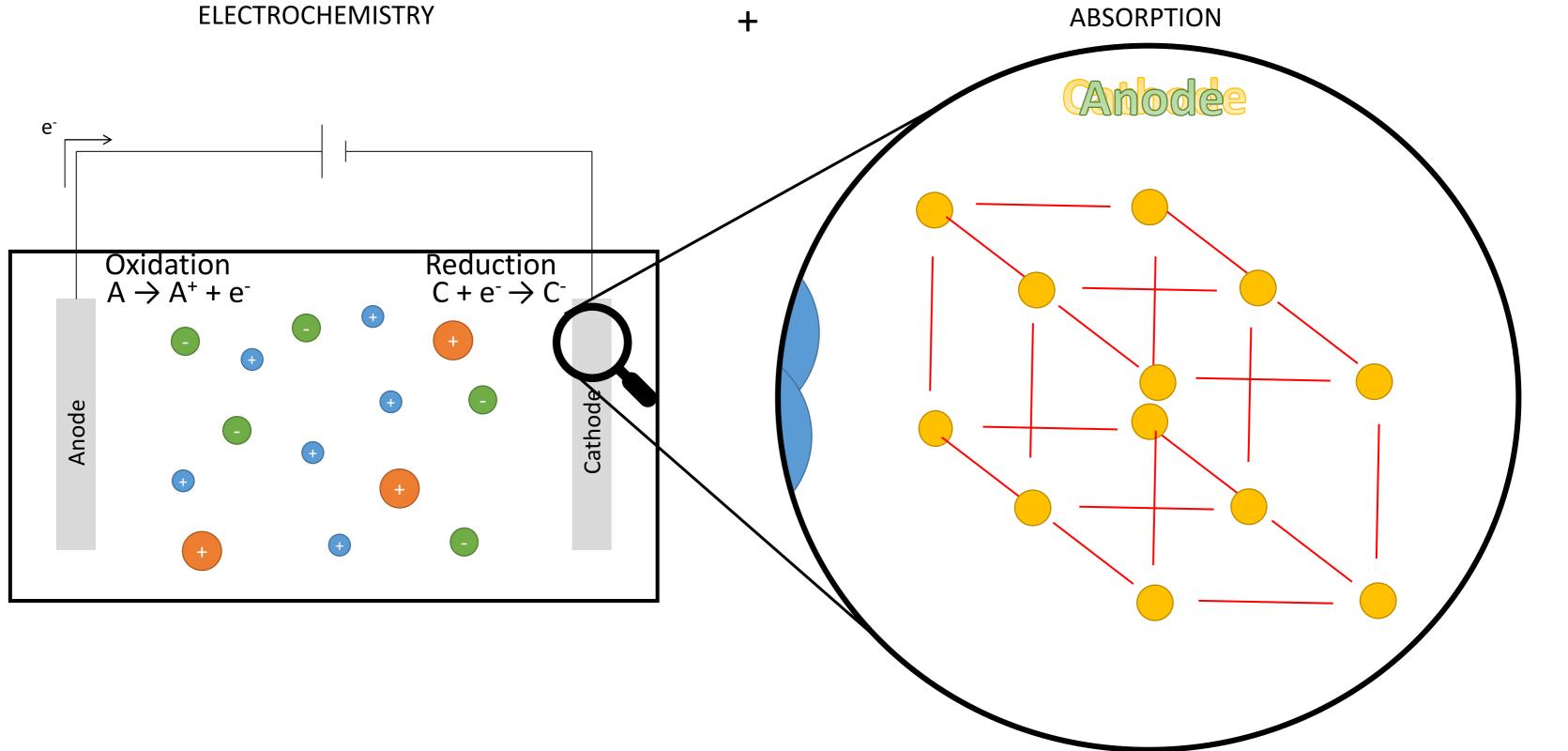


















Typical CDI cell operation:

• Electrodes of porous carbon.

+

- Low voltage (about 1.2V).
- Saline water in the cell.







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## 1. Sufficient surface area of activated carbon.

lons are adsorbed in the crystalline structure of the electrode:

The more material in contact with the solution — The more vacancies to capture ions.

2. Appropriate pore size distribution.

The size of the vacancies will determine the captured ions, therefore, the chosen crystalline depends on the target ion in the solution.







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Applications such as desalination of brackish water and seawater, water softening, removal of weak acids and **CRM recovery**.

Focus on electrode material suitable for this technology:

- Conductive.
- Selective with the target ion.
- High surface area.
- Cheap.
- Stable.







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Membrane Capacitive Deionization (MCDI) ullet

Monovalent Selective Membrane CDI (MSMCDI) ullet

Flow Electrode-Based Capacitive Deionization (FCDI) ullet





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Membrane Capacitive Deionization (MCDI): with Anion Exchange Membrane and ulletCation Exchange Membrane.





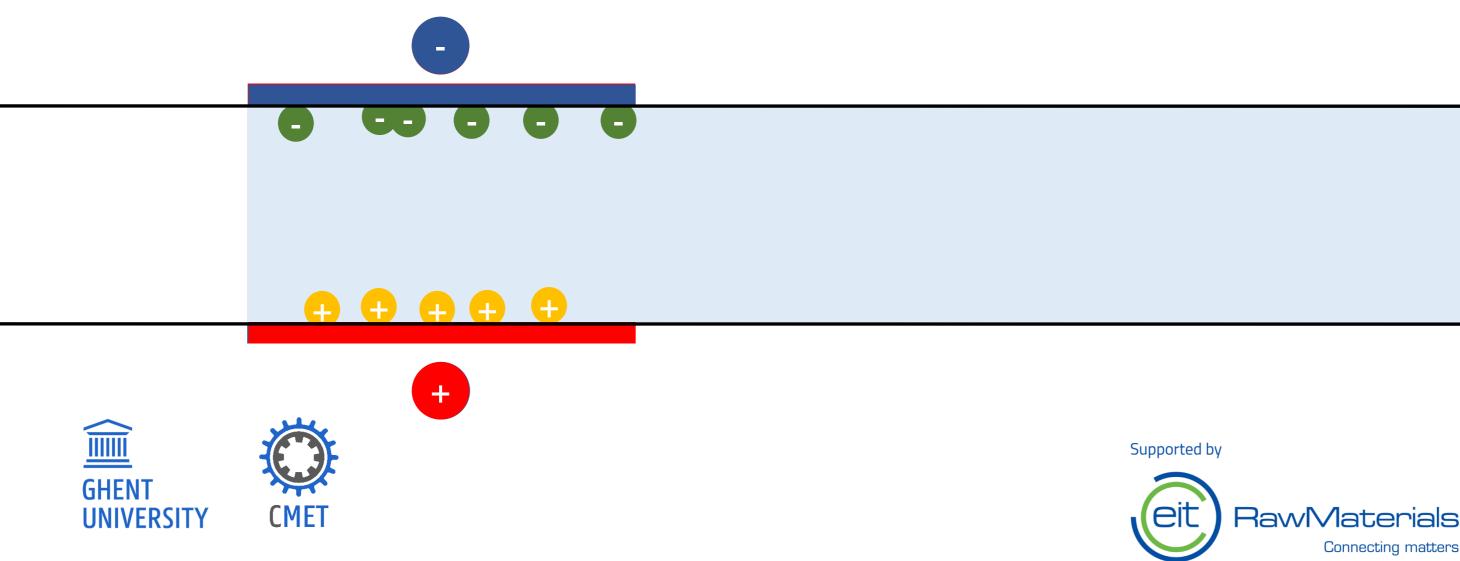
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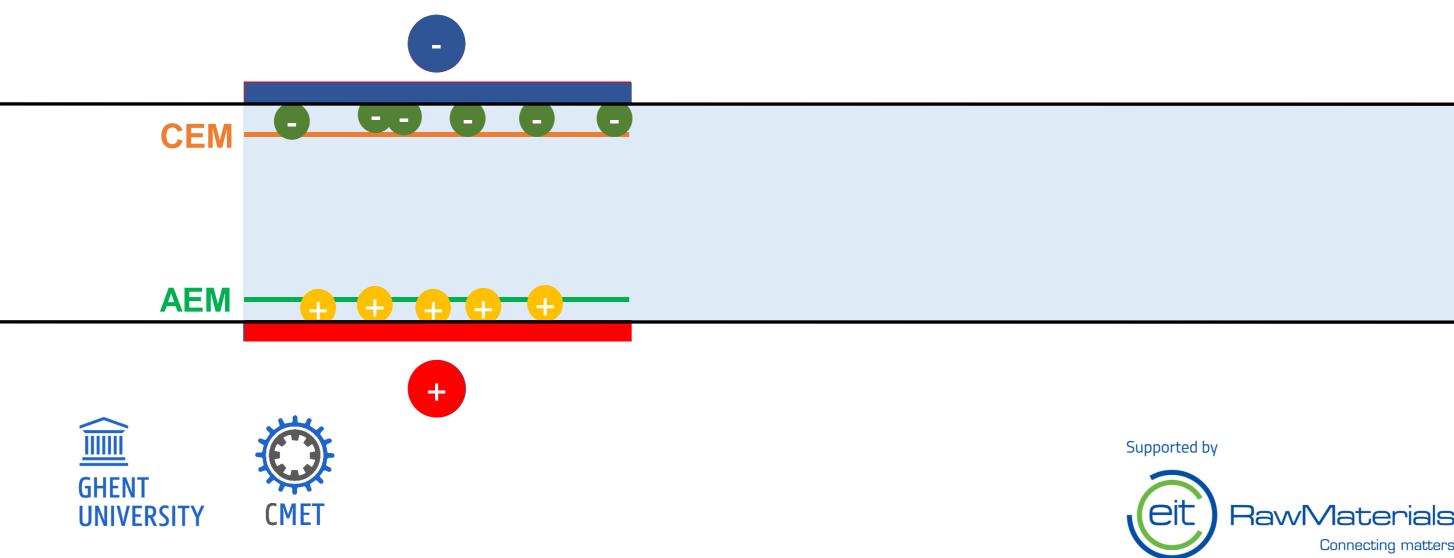
Normal Capacitive Deionization (CDI). ۲







Membrane Capacitive Deionization (MCDI): with Anion Exchange Membrane and Cation ulletExchange Membrane.







- Membrane Capacitive Deionization (MCDI): with Anion Exchange Membrane and Cation ulletExchange Membrane.
- Monovalent Selective Membrane CDI (MSMCDI): with Monovalent Cation Exchange ulletMembrane.





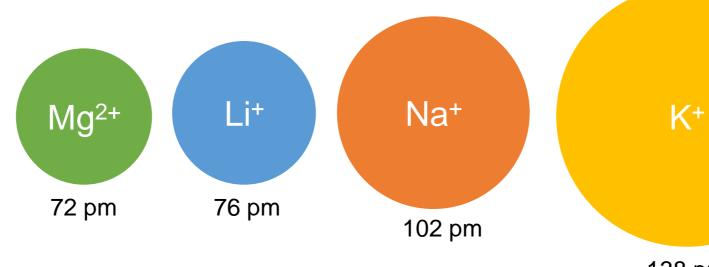
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- Membrane Capacitive Deionization (MCDI): with Anion Exchange Membrane and Cation ulletExchange Membrane.
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- Membrane Capacitive Deionization (MCDI): with Anion Exchange Membrane and Cation ulletExchange Membrane.
- Monovalent Selective Membrane CDI (MSMCDI): with Monovalent Cation Exchange ulletMembrane.
- Flow Electrode-Based Capacitive Deionization (FCDI): with suspended electrode ulletmaterial.





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Flow Electrode-Based Capacitive Deionization (FCDI): with suspended electrode ulletmaterial. ╋









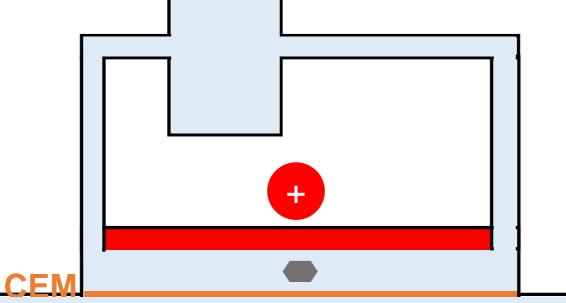
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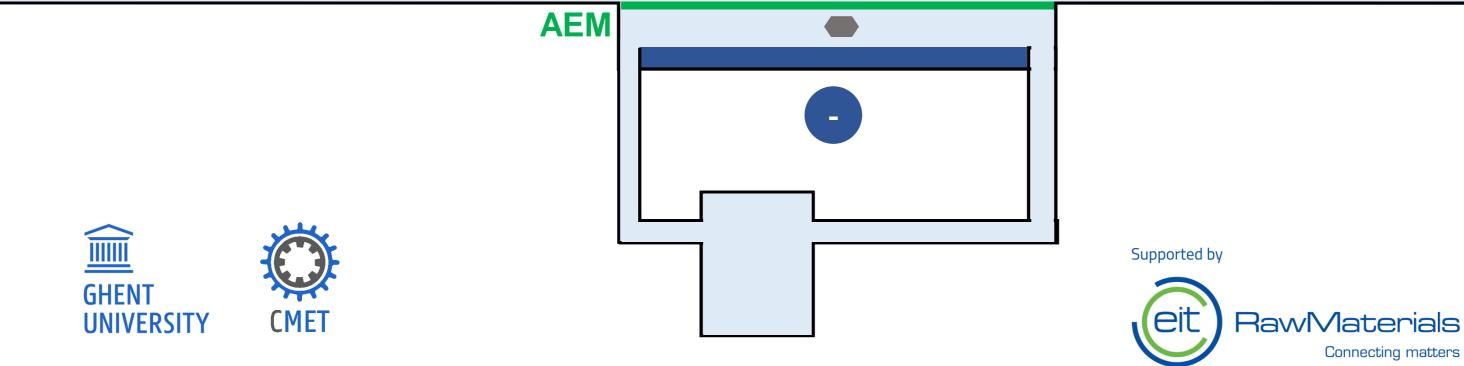






Flow Electrode-Based Capacitive Deionization (FCDI): with suspended electrode ulletmaterial.









- Arsenic ٠
- Copper  $\bullet$
- Vanadium  $\bullet$
- Lead  $\bullet$
- Chromium  $\bullet$
- Nickel  $\bullet$
- Cadmium  $\bullet$
- Lithium  $\bullet$
- Zinc  $\bullet$
- $\bullet$ . . .





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- Capacitive deionization is a technique to selectively remove ions from saline waters.  $\bullet$
- The removed ions can be recovered, what makes this technique interesting for CRM. ۲
- New electrode materials development is key for the selective capture of target ions.  $\bullet$
- The configuration of the capacitive deionization can help in the effectiveness and  $\bullet$ feasibility of the process.









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# Thanks for listening.

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